

## **6. BÖLÜM**

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### **ANYON AÇIKLI METABOLİK ASİDOZ**

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#### **Serum anyon açığına göre metabolik asidozun sınıflandırılması**

İnsan vücutundaki katyonların toplam sayısı, anyonların toplam sayısına eşit olmalıdır. Bu nedenle serum  $\text{HCO}_3^-$  konsantrasyonundaki düşüş bir diğer anyon konsantrasyonundaki artış ile dengelenir. Fazla hidrojen iyonlarına ( $\text{H}^+$ ) eşlik eden anyon  $\text{Cl}^-$  olduğu için, serum  $\text{HCO}_3^-$  konsantrasyonundaki azalma serum klor konsantrasyonundaki eşdeğer artış ile karşılaşır. Buna göre asidoz, hiperkloremik metabolik asidoz veya normal anyon açıklı metabolik asidoz olarak sınıflandırılır.

Fazla hidrojen, klor dışında bir anyon ile birlikte ise, bikarbonattaki azalma ölçülemeyen anyon konsantrasyonunda artış ile dengelenir. Klor konsantrasyonu aynı kalır. Bu durumda da asidoz normokloremik metabolik asidoz veya yüksek anyon açıklı metabolik asidoz olarak sınıflandırılır.

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